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## ELECTRONIC PROPERTIES OF SEMICONDUCTORS

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# Grain Boundary Related Electrical Transport in Al-rich $\text{Al}_x\text{Ga}_{1-x}\text{N}$ Layers Grown by Metal–Organic Chemical Vapor Deposition<sup>1</sup>

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**Abstract**—Electrical transport data for Al-rich AlGaN layers grown by metal–organic chemical vapor deposition (MOCVD) are presented and analyzed in the temperature range 135–300 K. The temperature dependence of electrical conductivity indicated that conductivity in the films was controlled by potential barriers caused by carrier depletion at grain boundaries in the material. The Seto's grain boundary model provided a complete framework for understanding of the conductivity behavior. Various electrical parameters of the present samples such as grain boundary potential, donor concentration, surface trap density, and Debye screening length were extracted.

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### 1. INTRODUCTION

In the recent years a considerable attention has been given to electrical properties of nitride compounds due to their significance in technology as well as in fundamental science. They have many applications for high electron mobility transistors (HEMT) and optoelectronic devices operating in the range from blue–green to ultraviolet [1]. Most of these applications are dependent on the remarkable quality of AlGaN. By the metal–organic chemical vapor deposition (MOCVD), high quality films of AlGaN/AlN can be grown on sapphire substrates [2]. On the other hand, we should consider that epitaxial layers of nitrides always have columnar structure depending on conditions of a growth process [3]. It was reported that the electron transport properties in GaN were strongly influenced by grain boundaries between ordered grains in the case of columnar microcrystalline growth of GaN [3–5].

In order to facilitate and to improve the operation of AlGaN based devices, a deeper understanding of the AlGaN electrical properties is demanded. However, when the carrier mobility in AlGaN is very small, the electron transport data are limited by the conductivity measurements. Therefore, electrical conductivity is a property of fundamental interest as well as of technological importance in such a case. The electri-

cal conduction mechanism of nitrides is sensitive to the crystalline nature of the structure (single/polycrystalline) [6, 7]. When the crystalline nature of the structure is considered, some valuable information related to the electrical properties can be obtained.

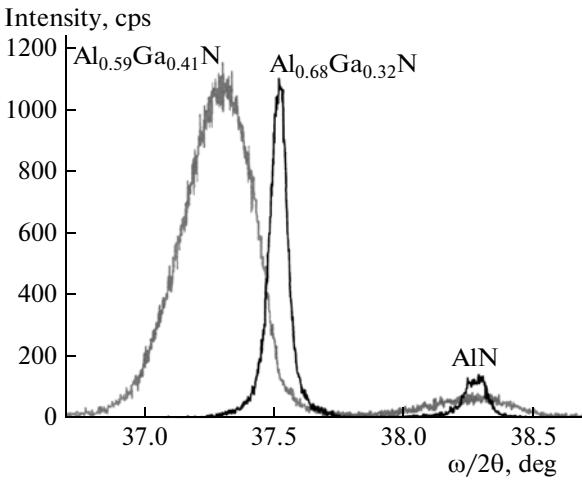
The electron transport properties of AlGaN single crystals were investigated in detail [8–10]. On the other hand, it is not possible to say the same things for polycrystalline AlGaN. In order to elucidate the fundamental significance of the electron transport within the polycrystalline AlGaN and, hopefully, to provide useful information about conduction in these structures, here we focus on the electrical conductivity of polycrystalline AlGaN layers. The structure of polycrystals plays an important role in the electrical properties of materials. Although, there are several works on grain boundary effects on electrical properties of GaN [3–6, 11, 12], to the best of our knowledge, data on the surface trap density of AlGaN layers have not been reported so far using temperature dependence of conductivity data. Therefore, it is important to determine this parameter for AlGaN based devices.

The main objective of this work is to highlight the related electrical properties of AlGaN layers having different grain sizes.

### 2. EXPERIMENTAL

The *n*-AlGaN/AlGaN/AlN structures were grown on *c*-plane [0001]  $\text{Al}_2\text{O}_3$  substrates in a low-pressure

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**Fig. 1.** HRXRD patterns of the AlGaN samples.

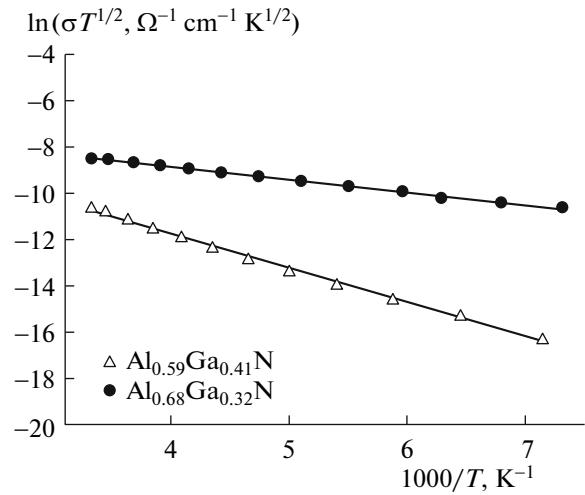
MOCVD reactor. Prior to epitaxial growth, the substrates were nitridated with 1000 seem  $\text{NH}_3$  flow at 790°C. At the same temperature, thin AlN nucleation layers were grown after nitridation. Then, ~150-nm high-temperature un-doped AlN buffer layers were grown at a temperature of 1075°C. After the AlN buffer layers, ~150-nm-thick undoped AlGaN layers were grown. Lastly, 480- and 400-nm-thick Si-doped *n*-type AlGaN layers were grown for Samples *A* and *B*, respectively. In AlGaN growth, the reactant source gas trimethylgallium (TMGa) flow was changed as 2 sccm in Sample *B* instead of 4 sccm in Sample *A*. All of the layers except the last AlGaN layer were nominally undoped.

Al mole fractions in the AlGaN layers were determined with a simple implementation of the Bragg's law using high-resolution X-ray diffraction (HRXRD) results. HRXRD measurements were taken with D8-Discover diffractometer equipped with a monochromator with four Ge(220) crystals for  $\text{Cu}K_{\alpha 1}$  X-ray beam (wavelength  $\lambda = 1.5406 \text{ \AA}$ ).

Conductivity measurements were taken with a Lakeshore Hall measurement system using  $5 \times 2 \text{ mm}$  van der Pauw samples in the temperature range  $T = 135\text{--}300 \text{ K}$ . Ohmic contacts for the conductivity measurements were prepared with evaporated triangular Ti/Al/Ni/Au (200/2000/300/700 Å) metals in the sample corners. After rapid thermal annealing ohmic behavior of the resulted contacts was confirmed by the current–voltage measurements.

Thickness ( $t$ ), the average grain size ( $L$ ), barrier height ( $E_b$ ), donor concentration ( $N_D$ ), Debay screening length ( $L_D$ ), and surface trap density ( $Q_s$ ) for the AlGaN samples

Sample	$t$ , nm	$L$ , nm	$E_b$ , meV	$N_D$ , $\text{cm}^{-3}$	$L_D$ , nm	$Q_s$ , $\text{cm}^{-2}$
$\text{Al}_{0.59}\text{Ga}_{0.41}\text{N}$	480	13.6	125	$3.44 \times 10^{18}$	2.18	$4.72 \times 10^{12}$
$\text{Al}_{0.68}\text{Ga}_{0.32}\text{N}$	400	88	49	$3.27 \times 10^{16}$	22.6	$2.88 \times 10^{11}$



**Fig. 2.** Temperature dependence of the conductivity plotted as  $\ln(\sigma T^{1/2})$  vs.  $10^3/T$ . Solid lines are the best-fit lines with Eq. (2).

### 3. RESULTS AND DISCUSSIONS

Figure 1 shows the high resolution close-up view of the diffraction peaks obtained using HRXRD measurements for the samples. In order to determine the aluminium content ( $x$ ) and the grain size ( $L$ ) in the samples, we scanned the (0004)  $\omega/2\theta$  reflections of hexagonal structure. From peak separations between AlGaN and AlN reflections, values of  $x$  were found using LEPTOS 4.02 with dynamic theory [13]. The values of  $L$  for the samples were calculated with the Debye–Scherrer formula [14]

$$L = \frac{0.9\lambda}{B\cos\theta}, \quad (1)$$

where  $B$  is the peak width,  $\theta$  is the diffraction angle and  $\lambda$  is the X-ray wavelength corresponding to  $\text{Cu}K_{\alpha}$  radiation. The values of the parameters  $x$ , and  $L$  are collected in table. The HRXRD measurements indicate that the film with  $x = 0.68$  have larger crystallites than the film with  $x = 0.59$  (table).

Figure 2 shows the plots of  $\ln(\sigma T^{1/2})$  ( $\sigma$  is the conductivity) vs.  $1000/T$  for the samples investigated. These data demonstrate that the conductivity increases with increase of grain size. This can be attributed to improvement in crystalline structure and leads to improvement in the conductivity. It is in agreement with Seto's grain boundary model [15] of

the conductivity which showed increase of conductivity with increasing grain size. A noticeable increase in conductivity with increasing grain size is observed in our case. The conductivity of the films increased from  $2.28 \times 10^{-5}$  to  $2.02 \times 10^{-4}$  ( $\Omega \text{ cm}$ ) $^{-1}$  at room temperature with the grain size increase from 13.6 to 88 nm, respectively. This can be elucidated on the basis of the Seto's grain boundary model. Then, it may be expected that the temperature dependence of the conductivity obeys the Seto's relation [15]

$$\sigma = \left( \frac{L e^2 n v_c}{k_B T} \right) \exp\left(-\frac{E_b}{k_B T}\right), \quad (2)$$

where  $e$  is the electron charge,  $n$  is the electron concentration in neutral region of crystallites,  $k_B$  is the Boltzmann constant,  $E_b$  is the barrier energy at the boundary and  $v_c$  is the collection velocity.  $E_b$  can be described as [15]

$$E_b = \frac{L^2 e^2 N_D}{8\epsilon}, \quad (3)$$

where  $\epsilon$  is the low frequency dielectric constant and  $N_D$  is the donor concentration.  $v_c$  is expressed as [15]

$$v_c = \left( \frac{k_B T}{2\pi m^*} \right)^{1/2}, \quad (4)$$

$m^*$  is the effective mass of charge carriers. By using the iteration method [16], the values of  $m^*$  and  $e$  of  $\text{Al}_x\text{Ga}_{1-x}\text{N}$  alloys as a function of  $x$  can be evaluated. Here, we used the values of effective mass  $m^* = 0.22m_0$  and  $0.48m_0$ , and the static dielectric constants  $\epsilon = 10.4$  and 8.5 for GaN and AlN, respectively [16]. Applicability of the grain boundary model involves many grain boundaries. This effect is examined by evaluation of the Debye screening length ( $L_D$ ) in comparison with  $L$ .  $L_D$  is given as [17]

$$L_D = \sqrt{k_B T \epsilon_0 \epsilon / e^2 N_D}, \quad (5)$$

where  $\epsilon_0$  is the dielectric constant of vacuum. If  $L_D < L/2$ , potential barriers exist in grain boundary region due to interface trap states [17]. If, however,  $L_D$  is larger than  $L/2$ , the conduction band becomes flat without the potential barrier [17], and the electrons are transported without grain boundary scattering.

Since a polycrystalline film has crystallites joined at their surfaces via grain boundaries, the boundaries between crystallites play an important role in determination of conductivity of polycrystalline films. In a polycrystalline material, high densities of defects are expected at grain boundaries which are often charged with majority carriers. The charged states at grain boundaries create depleted regions which also act as potential barriers [15, 17].

If we return to Fig. 2, the symbols in Fig. 2 are the experimental data and the solid lines are the best fitted

values with Eq. (2);  $r^2 = 0.99$  ( $r$  is the correlation coefficient) is obtained, which indicates the satisfactory fit. The linearity of the plots reveals that the grain boundary scattering of charge carriers is more predominant in the samples investigated. The potential barrier height in the films ( $E_b$ ) and the value of donor concentration ( $N_D$ ) were calculated from the slope of the curves in Fig. 2. The potential barrier height in the films decreased with increasing grain size. Decrease of the potential barrier height is due to increase of crystallite size resulting in diminishing charge carrier scattering at the grain boundaries. The shrinkage of a grain size leads to an increment in the trapping states at a grain boundary. Trapping states are capable of trapping free carriers and, as a consequence, more free carriers become immobilized as the density of trapping states increases. In other words, the larger grain size results in the lower density of grain boundaries, which behave as traps for free carriers and barriers for carrier transport in the film. Hence, increase in the grain size can cause decrease in grain boundary scattering, which leads to increase in the conductivity.

Knowing the values of  $N_D$ , the values of the Debye screening length ( $L_D$ ) can be calculated. The calculated values of  $L_D$  from Eq. (5) are given in table. Note that the condition  $L_D < L/2$ , appropriate for the grain boundary model is obeyed here for both samples. Thus, the approach of analyzing the data using the grain boundary model for thermal activation of conductivity is proper for both samples.

Charged states at the grain boundaries create the depleted regions and the potential barriers which provide resistance for the passage of carriers [15, 17]. This situation was also reported for nitrides [4–6]. Trapping states are capable to trap free carriers and, as a consequence, more free carriers become immobilized as number of trapping states increases. Now, the surface trap densities ( $Q_s$ ) in the films can be estimated using the relation [15]

$$Q_s = \frac{(8\epsilon\epsilon_0 N_D E_b)^{1/2}}{e}. \quad (6)$$

Substituting the values of  $N_D$  and  $E_b$  into Eq. (6), the values of  $Q_s$  are found and they are presented in table. The value of grain boundary surface trap density ( $Q_s$ ) should agree with the value of the surface state density of various systems having the same origin.  $Q_s$  is well in agreement with reported values for both nitride and other polycrystalline systems [15, 17, 18–20]. The values of  $Q_s$  decrease with increasing grain size and match the experimental data as expected [15]. Decrease in  $Q_s$  was also observed in AlGaN/GaN heterostructures after  $\text{Si}_3\text{N}_4$  passivation [18]. The improvement in the alignment of the grains at the grain boundaries minimizes the trapping of charge carriers at the grain boundaries.

The surface trap density ( $Q_s$ ) in Eq. (6) depends on the relation between  $N_D$  and  $E_b$ . It was reported that for

higher values of  $N_D$  in the grain inside, a higher  $Q_i$  is needed to form the potential barrier in GaN [4]. Here, this situation is also confirmed for our AlGaN layers.

#### 4. CONCLUSIONS

Electron transport data for Al-rich AlGaN layers grown by metal–organic chemical vapor deposition (MOCVD) are presented and analyzed in the temperature range 135–300 K. The data on temperature dependence of the conductivity were analyzed in terms of the grain boundary model. Characteristic grain boundary parameters, such as grain boundary potential, donor concentration, surface trap density, and Debye screening length, were all determined from our measurements where we show that their values quite well agree with the assumptions of the Seto's grain boundary model. It was found that the conductivity increases with increasing grain size. The potential barrier height and surface trap density decreased due to increase in grain size.

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